

High Pressure Investigations of Liquid and Polymerized CO up to 20 GPa Using Pair Distribution Function Analysis

Nadine Rademacher,^a Lkhamsuren Bayarjargal,^a Wolfgang Morgenroth,^a Jennifer Ciezak-Jenkins,^b Sasha Batyrev,^b Björn Winkler^a

^aInstitute of Geosciences, Goethe University, Frankfurt am Main, Germany, ^bUS Army Research Laboratory, Aberdeen Proving Grounds, USA. Email: Rademacher@kristall.uni-frankfurt.de

The analysis of the pair distribution function (PDF) is a powerful method for the determination of accurate structural parameters of crystalline as well as disordered, amorphous and nanocrystalline materials.^[1] *In situ* PDF studies at high pressure in diamond anvil cells (DAC) are challenging, because the set-up of DAC experiments in principle does not fulfill the requirements for high-quality PDF data, such as access to large scattering angles and high signal-to-background ratios.^[2]

The chemical reaction of carbon monoxide at high pressures has been studied extensively since a first report in 1983 and stability/phase diagrams have been proposed.^[3–6] Assumptions were made that polymeric CO (p-CO) consists of anhydride groups connected by linear chains and that the presence of chemical groups like ethers, esters and ketones is probable.^[6] However, structural information obtained from total scattering experiments is lacking. The diamond anvil cell (DAC) studies presented here have been performed with mixtures of 10–25 vol% CO in He in order to ensure quasi-hydrostatic conditions. During compression of the gas mixture CO separated from the He at around 3.6(2) GPa with He forming a large bubble within the liquid CO phase (see figure 1). Polymerization was induced photochemically at 4.8–5.8 GPa with a blue laser (473 nm, 80 mW) yielding a yellow to dark red solid (see figure 1). Total scattering experiments of liquid CO and p-CO up to 20 GPa were performed at the Extreme Conditions Beamline P02.2 at PETRA III (DESY, Hamburg) using 43 keV radiation.^[7] The PDFs of liquid CO and p-CO do not show structural correlations beyond 10 Å confirming the lack of long-range order in both phases (see figure 1). DFT-based atomistic model calculations revealed that the local atomic arrangement of liquid CO resembles the crystalline high pressure ϵ -CO phase. For the p-CO data analysis is still in progress, but a preliminary analysis shows the presence of interatomic distances characteristic of C–C and C–O bonds.

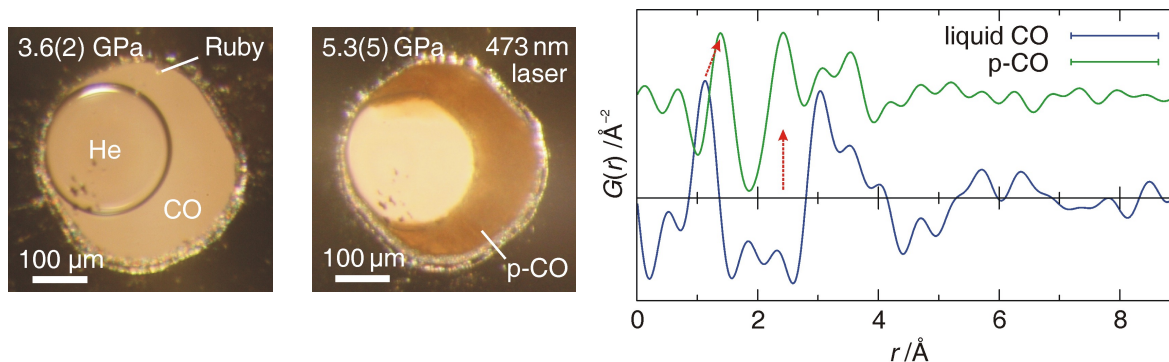


Figure 1: Left: Phase separation of CO and He in the DAC at 3.6(2) GPa. Middle: p-CO in He in the DAC at 5.3(5) GPa. Right: Experimental PDFs of liquid CO (blue) and p-CO at 0 GPa (green). The first-neighbour peak moves from 1.13 Å in liquid CO to around 1.4 Å in p-CO and a new peak appears at around 2.4 Å indicating the formation of the extended molecular solid p-CO.

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